

# Instrumentation of IR spectra

Theory:- Infrared Spectroscopy involves the interaction of IR radiation with matter. It covers a range of techniques, mostly based on absorption spectroscopy. The techniques can be used to identify and study the chemical substances. The sample may be solid, liquid, gas. The method or techniques of IR spectroscopy is conducted with an instrument called spectrophotometer to produce an IR spectrum. The IR portion of the electromagnetic spectrum is visually divided into three regions; Near-, mid-, far-IR spectrum. \*\*\* Core theory of IR, is discussed in vibrational spectroscopy.

Practical IR spectroscopy: The IR spectrum of a sample is recorded by passing a beam of IR light through the sample. When the frequency of a bond is the same as the vibrational frequency of a bond or collection of bonds, absorption occurs. Examination of the transmitted light reveals how much energy was absorbed at each frequency or wavelength. This measurement can be achieved by scanning the wavelength range using a monochromator. Alternatively, the entire wavelength is measured Fourier transform instrument and then a transmittance or absorbance spectrum is generated.

\* How would you obtain IR-spectra of a molecule practically?

## ① Sample preparation:-

① Gaseous samples require a sample cell with a long pathlength to compensate for the diluteness. A sample glass tube with length of 5 to 10 cm equipped with infrared transparent windows.

② Liquid sample can be sandwiched between two plates of a salt (commonly NaCl, CaF<sub>2</sub>, KBr). The plates are transparent to the IR light.

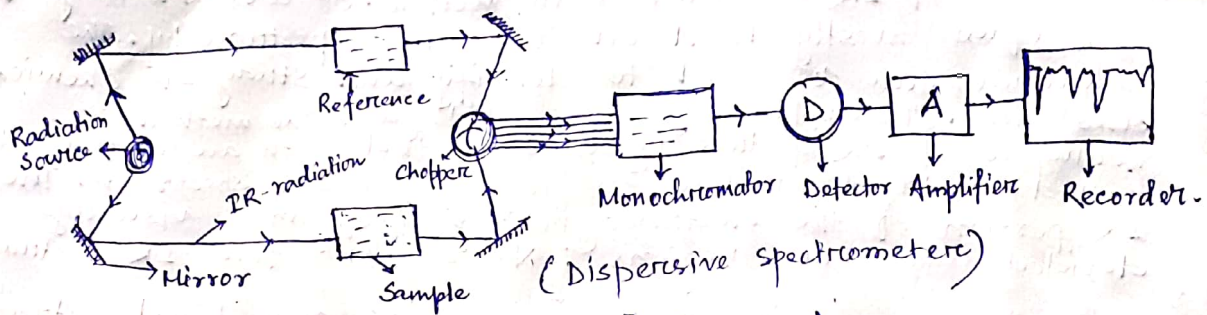
③ Solid samples can be prepared in a variety of ways. One common method is to crush the sample with an oily mulling agent (usually mineral oil Nujol). A thin film of the mull is applied onto salt plates and measured.

④ For polymeric solid sample 'Cast film' technique is used. The sample is first dissolved in a suitable non-hygroscopic solvent. A drop of this solution is deposited on surface of KBr or NaCl cell. The solution is then evaporated to dryness and the film formed on the is analysed directly.



# Photodetectors

## Instrumentation



Instrumental Setup for IR-spectrum.

\* Write down the characteristics of any one Radiation source for IR-spectra?

### (Radiation source)

- Nernst Glower: i) Oxides of Zr, Y, Er
  - ii) A hollow rod, diameter - 2mm, length 30mm
  - iii) Non conducting at room temp, heating required at (1650 - 1850°C)
  - iv) Produce wide IR range and intense spectra.

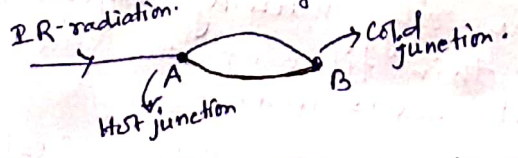
- Globar: i) Rod of sintered Silicon Carbide
  - ii) Diameter is 4mm & length 50mm.
  - iii) Heating required to produce IR-spectrum at 1300 - 1700°C.
  - iv) less intense-IR radiation produce.

\* In reference generally KBr is used as, because it can transmit all IR radiation.

- Monochromator: → ① Prism (Basically NaCl prism is used)  
 ② Gratings

Detector: ① Bolometer: a) It consist a thin metal conductor.  
 b) With increasing temp, resistance of metal conductor increases hence current flow changes which record by recorder.

② Thermocouple: Due to temp. difference potential difference will produce. Hence current flow changes.

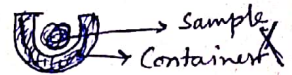


(iii) Thermistor: fused mixture of metal oxides are used. Due to IR-radiation temp. of metal oxides are ~~change~~ increases as a result resistance will decrease.

(iv) Golay detector

(v) photoconductivity detector.

1. Solid Sampling techniques: (i) Direct sampling.



(ii) Pelletization techniques: - • KBr } used for preparation of  
NaCl } Pellet with sample.

• The alkyl halide are able to pass the IR radiation throughout the IR-range that is why KBr, NaCl are used for this technique.

• In this technique sample is properly mixed with KBr or NaCl and the particle size should be less than  $2 \mu\text{m}$ . The mixture is kept under two pistons to form a pellet.

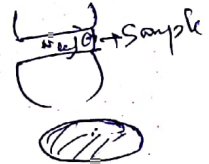
• Only NaCl or KBr Pellet used as reference.

(iii) Mulling technique: It is also known as paste technique. (Paste/Mull). Basically mulling agent i.e. mineral oil Nujol is used.

Mulling agent:- Nujol, Hexachlorobutadiene, CFC.

• Nujol has a) C-H stretching <sup>frequency</sup> at  $3030-2860 \text{ cm}^{-1}$

b) C-H bending  $1460-1374 \text{ cm}^{-1}$



(iv) Liquid Sampling technique:

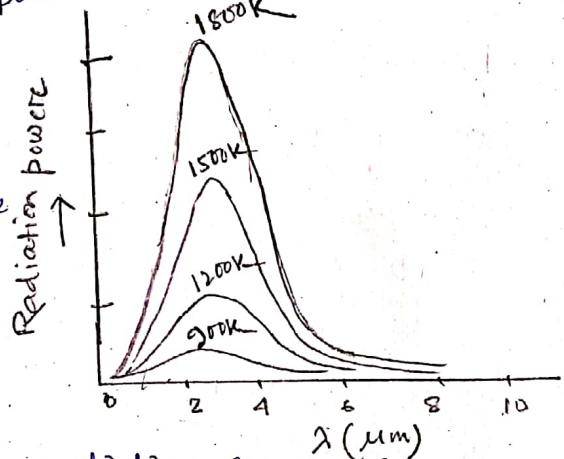
\* Q: Discuss the sampling technique in IR spectrometer.



## Radiation Sources:-

The essential element for absorption spectroscopy is a radiation source which show an as high as possible intensity in the wavelength area under the investigation. Thermal radiator are mainly used which provide a very broad band, so called continuum radiation. In contrast the use of lasers for measuring absorption spectra presumes the use of monochromatic radiation that have an extremely narrow band.

In IR spectroscopy, Planck radiator are used as radiation sources for sample excitation. The intensity of radiation emitted by a black radiator is subjected to the plank radiation principle. According to this principle, the emitted spectral radiation power reaches a maximum depending on temp. On the short wavelength side the radiation power reaches maximum and the curve falls steeply and towards high wavelength region the curve drops off flatter. Depending on the spectral region of interest, whether it be the far, middle or near IR, different sources are used.



The most frequently used radiation source in the mid-IR spectral region is the Globar. It consist by SiC (silicon carbide) in the forms of rod or helixes. As the result of its electrical conductivity in a cold state, the Globar can be directly ignited. At a burning temp. of about 1500K, its power uptake is quite substantial, so that in most cases the sources casing in the spectrometer is equipped with water cooling. The use of Globar has an advantage that, its emissivity is relatively high to about  $100 \text{ cm}^{-1}$ , thereby allowing it also to be used as a sources in the far-IR region.

## Q: Advantages of use of Nernst Rod radiator Over Globar:

In comparison, the Nernst Rod is used as a radiation sources. Because -

- (i) It has higher working temp. than Globar.
- (ii) One rod is relatively sensitive, mechanically speaking and particularly upon improper mounting is prone to deformation possibly impairing the optical stability of spectrometer.
- (iii) It made of  $\text{ZrO}_2$  with additives  $\text{Y}_2\text{O}_3$  (Yttrium oxide) which as a -ve temp. coefficient of electrical conductivity resistance.
- (iv) It is a non conductor at room temp. hence an initial auxiliary heating is required for ignition.
- (v) The normal operation temp. lies at about 1900K so that its emission maximum appears at between 1 and 2  $\mu\text{m}$ .



Q: Characteristics of spectrometers used for Near-IR? —

- (i) The spectrometers contain a source which is made by metallic helices mainly - Cr-Ni alloys (Tungsten).
- (ii) For the near-IR region tungsten-halogen lamps are used exclusively which allow a higher operating temp, thus a higher radiation is produced.

• For the far-IR region, particularly for the measurements below  $100\text{ cm}^{-1}$ , a mercury high pressure lamp is suited.

Q: Write down the types of detector for detecting the IR-radiation?

Infrared Detectors:— Before the principles of spectral resolution, the common radiation detector for IR-spectroscopy should be introduced. Their job is to convert the optical signal into further applicable electrical signal. There are several detectors.—

- i) Thermal detectors
- ii) Pyroelectric detectors
- iii) Pneumatic detectors.
- iv) Photoelectric detectors.

Besides the detection, lifespan, other criteria for evaluating detectors is the applicable wavelength range as well as the sensitivity, represented by the change of measurements signal as a function of the change of the radiation power. Moreover, important is the signal to noise ratio (SNR) and the time constant of the response velocity.

Q: Define specific detectivity?  
 A characteristic parameter, the specific detectivity  $D^*$ , has been shown useful for comparing different detectors with one another —

$$D^* = F_D^{1/2} / \text{NEP}$$

$$\text{Unit of } D^* = \text{cm} \cdot \text{Hz}^{1/2} \text{ W}^{-1}$$

$$F_D = \text{Detector area in cm}^2$$

\* Define NEP and state its unit.  
 NEP = Noise equivalent power in  $\text{W Hz}^{-1/2}$

NEP:— It is the incident radiation power on the detector that leads to a signal-to-noise ratio of one at the available electric bandwidth. This power can be defined and measured by the following equation—

$$\text{NEP} = \frac{\Phi}{(4f^{1/2}, S/N)}$$

$\Phi$  = Radiation power in W.  
 $4f$  = electric bandwidth in  $\text{Hz}$ .  
 $S/N$  = Signal to noise ratio.



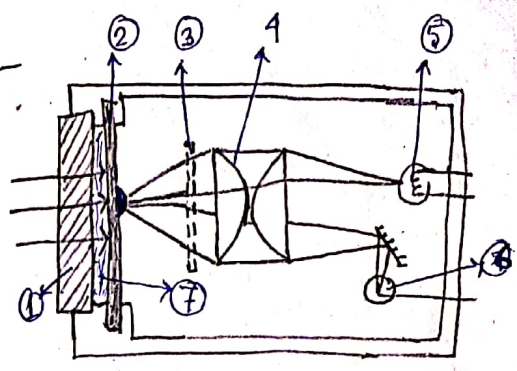
Q: Write down the basic principle of Golay detector? Write down the advantages of it?

Thermal Detectors :-

Q: Basic principle of Golay detector :-

Golay detector belongs to a subgroup of thermal detectors. The principle of Golay detector can be describe as follows. -

Radiation enters through an IR-permeable window into a gas filled cell and is absorbed there on a blackened film. The absorption heat causes an increase of a gas pressure to which the back wall of the cell

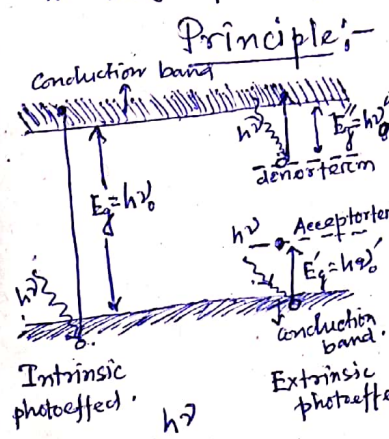


- Golay detector.
- ① IR-Permeable window
  - ② Blackened membrane with a reflective coating on the backside
  - ③ Grating    ④ Lens
  - ⑤ Light source    ⑥ Photodiode.

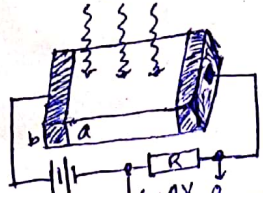
It consists of a gas filled enclosure (7) with an infrared absorbing material and a flexible diaphragm or membrane. When IR radiation absorbed, it heats the gas, causing it to expand. The gas pressure will be increased which deforms the membrane. Light reflected off the membrane is detected by photodiode and motion of the membrane produces a change in the signal on the photodiode.

- ✓ Advantage :-
- (i) The Golay Cell has high sensitivity and a flat response over a very broad range of frequency.
  - (ii) The response time is modest, of order 10 ms.
  - (iii) It must be protected from the foreign light by an automatic lid.

Photodetectors :- Electromagnetic radiation can interact in many ways with the material. Because of their higher sensitivity, photoelectric detectors are mostly used in IR spectrophotometers.



Principle :- Incident radiation alters the electrical conductivity in the irradiated semiconductor material. The photosignal is measured either as a change of voltage via the resistance R or as change in current. The elementary process of photoconduction is always the production of an electron-hole pair, whereby either both charge carriers are free to move in the electric field (intrinsic process) or one of the two charge carriers is spatially bound (extrinsic process).





Depending upon spectral apparatus, there are two types of spectrometers —

- ① Dispersive spectrometer
- ② Non-dispersive spectrometer.

Dispersive spectrometer:— This is generally double beam, recording instruments in which a reflecting grating is used for dispersing the IR radiation. Generally dispersive IR spectrometers incorporate a low frequency chopper that permits the detector to discriminate between the signal from the source and signal from extraneous radiation.

Non-dispersive:—

In non-dispersive spectrometer no variable wavelength selection is possible. It is often used to detect gas. No dispersive elements like prism, diffraction grating are used in this spectrometer. It measures the concentration of  $\text{CO}_2$ ,  $\text{CO}$ .

Principle of Non-dispersive IR:—

An infra-red beam passes through the sampling chamber and each gas component in the sample chamber absorbs some particular IR frequency. In parallel a reference gas, typically  $\text{N}_2$  is used in another chamber. By measuring the amount of absorbed IR at the necessary frequency, the concentration of the gas component can be measured. It is called non-dispersive because the wavelength which passes through the sampling chamber is not pre-filtered.

Disadvantage of non-dispersive IR:— Due to absence of optical filter in front of detector the sample can not absorb monochromatic radiation. As a result we do not obtain exact data which we expect.

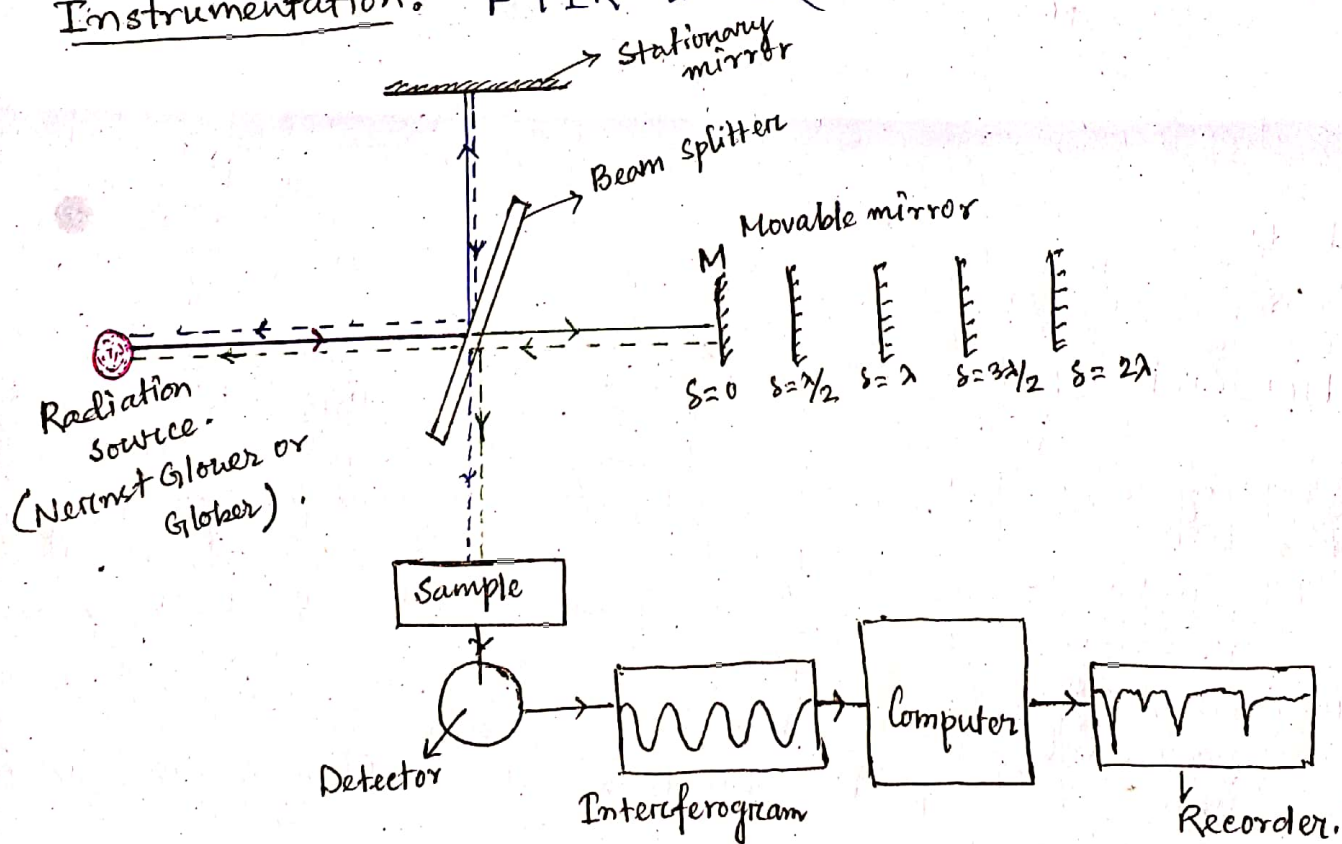
## Fourier-transform IR Spectroscopy

FTIR - is a technique used to obtain an IR spectrum of absorption or emission of a solid, liquid or gas. An FTIR spectrometer simultaneously collect high spectral resolution data over a wide spectral range. This confers a significant advantage over a dispersive spectrometer, which measures intensity over a narrow range of wavelengths at a time.

Principle: FTIR is based on the principle of interferometry and interferogram will be obtained, which is a complex signal occurs in wavelike pattern.

Interferogram signal is plotted wave intensity vs time. From this plot a mathematical operation is carried out called Fourier transformation. The spectra is called FTIR.

Instrumentation: FTIR is ~~also~~ also known as interferometer.



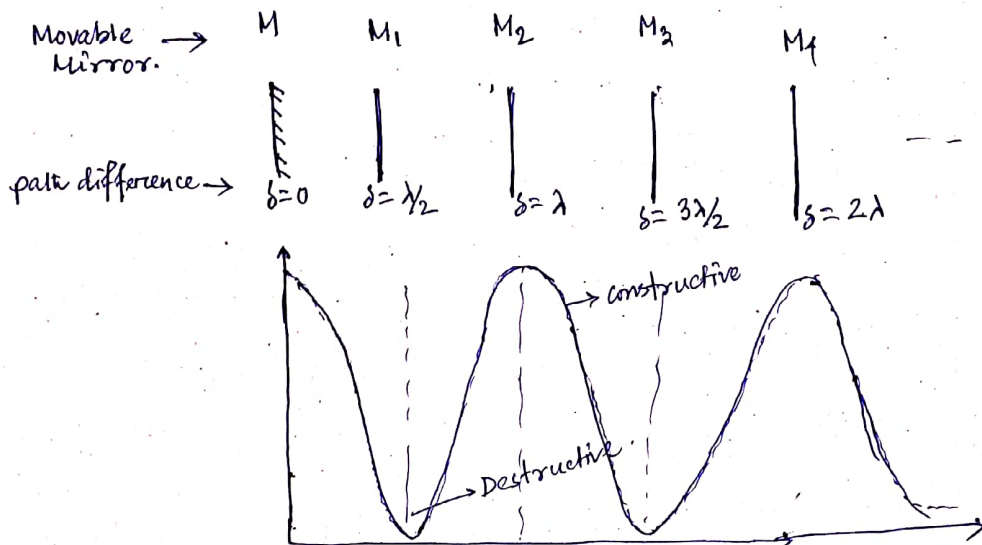
Block diagram of FTIR



Working procedure: - In FT-IR Spectrometer, Nernst Glower or Globar is used as radiation source.

The radiation is half-half transmitted through beam-splitter. 50% radiation goes into stationary mirror and another 50% goes into movable mirror. Two transmitted radiation goes into sample tube. From sample tube another radiation is produced which is detected by thermal detector.

This detector convert the light signal into electrical signal called interferogram. This interferogram is treated for converted into spectrum by Fourier transformation in the computer. And finally, a wave intensity vs time plot is shown in recorder. We will get the FT-IR spectra.



This interferogram is divided into two parts -

(i) constructive signal: If  $\delta$  is integral multiple of  $\lambda$  [0, 1, 2, ...]

(ii) Destructive signal: If  $\delta$  is half-integral multiple of  $\lambda$ .

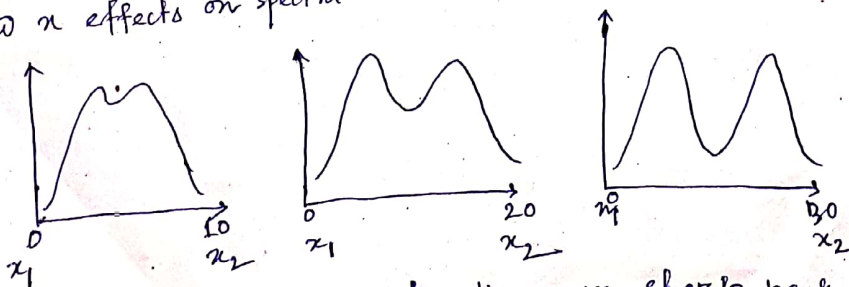
\* This interferogram is treated in Fourier transformation

Ex:  $y = a_0 \sin(\omega x) + b_0 \cos(\omega x) + a_1 \sin(1\omega x) + b_1 \cos(1\omega x) + \dots + a_n \sin(n\omega x) + b_n \cos(n\omega x)$

$$y = \sum_{n=0}^{\infty} a_n \sin(n\omega x) + b_n \cos(n\omega x)$$

Now:  $\omega = \frac{2\pi}{(x_2 - x_1)}$  where  $(x_2 - x_1)$  is the scale of  $x$  axis

⊕ How  $x$  effects on spectra.



i.e. With increase the scale the more sharp peak will obtained.

## Advantages of FT-IR:-

- (i) The main advantage is that S/n ratio is very low with compared to IR.
  - (ii) It has better resolution power.
  - (iii) High scanning speed and multiple scanning.
  - (iv) By using mathematical eqn. ~~it~~ it will give better result with compared to IR.
  - (v) It has better accuracy than IR.
- o —